# COAL-SPOIL AND GROUND-WATER CHEMICAL DATA FROM TWO COAL MINES; HANNA BASIN AND POWDER RIVER BASIN, WYOMING

By L. R. LARSON

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#### CONVERSION FACTORS

For use of readers who prefer to use metric (International System) units, rather than the inch-pound units, the following conversion factors may be used:

Multiply	<u>By</u>	<u>To obtain</u>
foot (ft)	0.3048	meter
inch	25.40	millimeter
mile	1.609	kilometer
ton, short	0.9072	tonne

Temperature in degrees Fahrenheit (°F) can be converted to degrees Celsius (°C) as follows:

$$^{\circ}C = 5/9(^{\circ}F - 32^{\circ})$$

<u>Sea level</u>: In this report "sea level" refers to the National Geodetic Vertical Datum of 1929 (NDVD of 1929)—a geodetic datum derived from a general adjustment of the first-order level nets of both the United States and Canada, formerly call "Mean Sea Level of 1929."

# COAL-SPOIL AND GROUND-WATER CHEMICAL DATA FROM TWO COAL MINES; HANNA BASIN AND POWDER RIVER BASIN, WYOMING

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#### ABSTRACT

Coal-spoil data and water-quality data for samples collected from two Wyoming coal-mining areas are presented in six tables. The Medicine Bow and Seminoe Number 1 Mines in the Hanna basin comprise one site, and the Cordero Mine in the Powder River basin is the other site. The data consist of X-ray diffraction analysis of spoil samples, analysis of the saturated paste extract of spoil samples, sulfur-forms analysis of spoil samples, analysis of water samples obtained from batch-mixing experiments, and analysis of ground-water samples collected from wells in the spoil and coal aquifers. The data were collected as part of a regional study which will evaluate methods used to predict post-mining ground-water quality of aquifers affected by the strip mining of coal in the West.

#### INTRODUCTION

#### Purpose and Scope

A considerable amount of data has been collected from mines in Wyoming. Montana, Colorado, Oklahoma, North Dakota, and Texas in order to evaluate methods used to predict post-mining ground-water quality of aquifers affected by the strip mining of coal in the West and to determine the geochemical processes involved. The purpose of this report is to document all data collected in Wyoming for this regional study. The data collected 1. mineral composition of spoil samples, 2. are as follows: paste-extract analysis of spoil samples, 3. sulfur-forms analysis of spoil samples, 4. chemical analysis of water samples obtained from batch-mixing experiments, 5. common-constituent analysis of water samples from coal and spoil aquifers, and 6. trace-constituent analysis of water samples from coal and spoil aquifers. The two Wyoming mining locations selected for inclusion in the regional study are the Medicine Bow-Seminoe Number 1 Mines in the Hanna basin and the Cordero Mine in the Powder River basin. (The adjoining Medicine Bow and Seminoe Number 1 Mines are treated as a single site in this report.) Spoil samples and ground-water samples from the coal and the mine spoil were collected in 1984 and 1985.

#### Physical and Climatic Settings

The Medicine Bow-Seminoe Number 1 Mines are in Carbon County, northeast of Rawlins. The Medicine Bow Mine is located between the two arms of Seminoe Reservoir which are formed by the North Platte River and the Medicine Bow River (fig. 1). Seminoe Number 1 Mine is south of and contiguous to the Medicine Bow Mine. These two adjacent strip mines are on the western side of the Hanna structural basin. The Ferris Formation of Paleocene age, which outcrops here, has seven coal seams which have been or

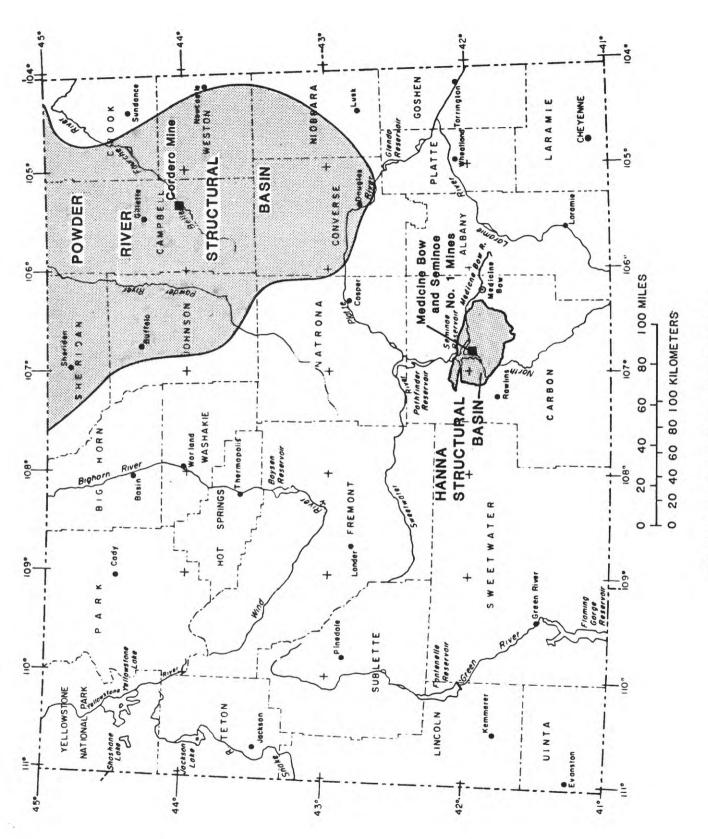


Figure 1.--Location of mine sites.

are being mined. Several other mines, both active and inactive, are located to the east of the Medicine Bow-Seminoe Number 1 Mines. The coal seams in the Ferris Formation at Medicine Bow-Seminoe Number 1 Mines dip about 6 degrees eastward-toward the center of the basin.

Precipitation at Medicine Bow-Seminoe Number 1 Mines averages about 11 inches per year, although significant variations occur from year to year. Annual precipitation at Rawlins, about 20 miles to the southwest, ranged from about 6.5 inches to 11.3 inches during the 15 years from 1966 to 1980. Annual precipitation at the town of Medicine Bow, about 30 miles east of the mines, ranged from 7.4 inches to 19 inches for the same 15-year period (U.S. Department of Commerce, 1984).

The Cordero Mine, in the coal-rich Powder River structural basin, is located about 20 miles south of Gillette in Campbell County (fig. 1). More than 88 million tons of coal were mined in Campbell County during 1983 (Jones, 1984, p. 32). The Cordero Mine, which began operation in 1976, is one of about a dozen mines presently extracting a portion of the thick layer of strippable coal that extends diagonally from the northwest corner of Campbell County to the southeast corner of the county and beyond. The Wyodak-Anderson coal and stratigraphically equivalent beds in the Fort Union Formation of Paleocene age are the source of the coal being mined.

The overburden at the Cordero Mine consists of rock from the Fort Union Formation and the overlying Wasatch Formation of Eocene age. As mining progresses, the dip of the coal seam and the increase in the altitude of the land surface results in an increasing proportion of rock from the Wasatch. Connor and others (1976) have pointed out that the mineralogy of these formations differs one from another.

Precipitation at the Cordero Mine averages about 13 inches per year, based upon data collected at the mine. More than half of the annual precipitation usually falls during April, May, and June. Rainfall, however, is highly variable, as indicated by the range in values for precipitation at Gillette for the month of May. The lowest rainfall measurement on record for May is 0 inches—the highest is more than 11 inches. High summer evapotranspiration rates on the prairies, much in excess of normal summer rainfall, usually result in the drying of prairie grasses by early summer at both the Powder River and Hanna basin mine sites.

The ranges of average daily temperature at Rawlins and Gillette are indicated in the following table:

	Average dail	y temperatur	es, in degre	es Fahrenheit
	Ju	ly	Jan	uary
City	maximum	minimum	maximum	minimum
Rawlins Gillette	84 88	51 56	31 33	12 11

The relatively high altitude of the mine areas moderates the summer temperatures. The altitude of Medicine Bow-Seminoe Number 1 Mines is about 6,500 feet above sea level, and the altitude of Cordero Mine is about 4,500 feet. The normally clear skies and dry air permit rapid radiation of heat after sunset, which results in cool evenings, even during the summer months (Martner, 1981, p. 5-8).

#### **Acknowledgments**

The cooperation and assistance of personnel from the Arch Minerals Corporation and the Sundeco Coal Company are greatly appreciated.

#### METHODOLOGY

#### Spoil Samples

Spoil samples were collected at three sites at each of the two mine sites. Three samples were collected at each site: one near the base of the spoil, one near the top of the spoil, and one midway between the base and the top. In addition, duplicate samples were collected at one of the three sites to provide a basis for determining repeatability of sampling technique and laboratory analyses. Thus, a total of 12 spoils samples were collected from each mine site.

The spoil-sample designation consists of four or five characters. The first character, a letter, designates the state (W = Wyoming); the second, a number, designates the mine site (1 = the Medicine Bow-Seminoe Number 1 Mines, 2 = the Cordero Mine); the next character, a letter, designates the sampling site; and the fourth character, a number, designates the vertical location at a site (1 = base, 2 = midway, and 3 = top). When used, the fifth character, the letter D, designates a duplicate sample of spoil material was taken and analyzed. The letter S (used in the batch-mixing experiments analysis) indicates the water sample was split for analysis. (The split samples give an indication of laboratory precision.)

Each spoil sample was air dried, thoroughly mixed, and then divided. Part of each sample was sent to the U.S. Geological Survey (Geologic Division) Laboratory, Denver, Colorado for crushing and grinding. The mineral composition (table 1) was determined (Schultz, 1960; Jacobs, 1984) by the North Dakota Mining and Mineral Resources Research Institute, Grand Forks, North Dakota. Analysis (table 2) of the saturated paste-extract (Sandoval and Power, 1977) was by Minnesota Valley Testing Laboratories, Inc., Bismarck, North Dakota. The sulfur-forms analysis (table 3) was determined (Jackson and others, 1987) by the U.S. Geological Survey (Geologic Division) Laboratory, Denver, Colorado.

The spoil material that was not crushed and ground was used in batch-mixing experiments using a method as described by Davis (1984). In these experiments, water and spoil were combined in weight ratios of 2:1 and 5:1 (water:spoil). Sample designations (table 4) with ending numbers 21, have a 2:1 ratio; those with ending numbers 51, have a 5:1 ratio. The water used in the experiments was obtained from a well in the coal at each site in order to simulate the assumed major source of recharge to the spoil. Mixing

was accomplished by fastening the bottles (which were completely filled with the water and spoil mixture to exclude oxygen) to a vertical wheel 3 feet in diameter and then rotating them for 2 hours at 3 revolutions per minute. The spoil was then allowed to settle. After a total contact time of about 24 hours, the water was filtered first through a 0.45-micrometer filter and then through a 0.10-micrometer filter. The filtrate was analyzed (table 4) by the Geological Survey (Water Resources Division) Laboratory, Denver, Colorado.

### Water Samples

Water samples were collected from wells completed in the coal and wells completed in the spoil at each mine site. The samples were analyzed in the field for specific conductance, pH, temperature, and bicarbonate and car-If the odor of hydrogen sulfide was detected, the bonate concentrations. samples were analyzed at the sampling site for sulfide. If the hydrogen sulfide odor was not detected, the samples were analyzed on site for The samples also were analyzed (Skougstan and others, dissolved oxygen. 1979) by the Geological Survey (Water Resources Division) Laboratory, Denver, Colorado for selected common and trace constituents. The water samples were filtered through a 0.45-micrometer filter. In addition to the 0.45-micrometer filtration, water analyzed for silica, aluminum, iron, and manganese was filtered through a 0.10-micrometer filter. Although filtration through a 0.45-micrometer filter generally is assumed to pass only dissolved constituents, a 0.1-micrometer filter was used to ensure that colloidal forms of silicon, aluminum, iron, and manganese did not pass through the filter. This step is critical in geochemical studies.

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DATA

page 4 follows

Table 1.--Mineral composition of spoil samples determined by X-ray diffraction

[Analyses by the North Dakota Mining and Mineral Resources Research Institute. Concentrations in percent by weight. Abbreviation: <, less than. Sample numbers ending in the letter "D" designate a duplicate sample]

Sample	Mixed- layer clays	Smectite	Chlorite	Muscovite 1111te	Kaolinite	Gypsum	Quartz	Potassium feldspar	Plagio- clase feldspar	Dolomite	Calcite	Total
				Med	Medicine Bow-Seminoe		Number 1	Mines				
W1A1 W1A2 W1A3	510	404	409	10 7 8	24 28 28	Q W 4	26 37 35	-10E	2002	Û40	040	999 943
W1B1 W1B1D	77	ro 4	4 ro	တထ	29 31	14	32 24	04	04		00	102 97
W1B2 W1B2D	ლ გ	40	ოდ	<b>ω</b> ω	29 35	ოო	33 28	4 00	დო	22	m 72	86 66
W183 W183D	10	00	12	15 11	50 20	e	30	40	73	175	12 17	101 96
W1C1 W1C2 W1C3	0 1 5	m N O	<b>∞</b> ⊶4	4 8 11	30 9 33 9	16 0	26 45 36	W/4	8 0 0 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	m 0.4	10 2	95 97 101
					3	Cordero Mi	Mine					
W2A1 W2A2 W2A3	ဖတဝ	800	010	W 4 L	37 18 44	13 16	31 33 28	4 & &	<u>1</u> ,22	Trace 2 <1	000	105 98 100
W2B1 W2B1D	13	26 24	88	11	15 19	41	24 28	52	00	Trace 0	00	98 100
W2B2 W2B2D	13	21 16	ოო	ထထ	23 38	00	30	17	00	00	00	100
W2B3 W2B3D	10 9	13 14	00	<b>4</b> W	∞4	w <b>4</b>	45 51	19	00	7	77	105 103
W2C1 W2C2 W2C3	5 7 9	36 18	ខ្លួក	യവവ	31 16 32	110	38 27 31	222	000	<1 Trace 0	0 7 0	95 102 100

Table 2.--Analysis of the saturated paste extract of spoil samples

[Analysis by Minnesota Valley Testing Laboratories, Inc. Concentrations in milligrams per liter (mg/L) or milliequivalent per 100 grams (meq/100 g); specific conductance in microsiemens per centimeter at 25 degrees Celsius ( $\mu$ S/cm). Sample numbers ending in the letter "D" designate a duplicate sample]

		Specific	Ion co	ncentration extra		ration	Cation exchange
Sample_	Paste pH	conduc- tance (µS/cm)	Calcium (mg/L)	Magnesium (mg/L)	Sodium (mg/L)	Sulfate (mg/L)	capacity using ammonium acetate (meq/100 g)
		М	edicine B	ow-Seminoe	Number 1	Mines	
W1A1	5.30	6,653	92	51	17	73	5.65
W1A2	6.80	5,363	154	27	6.1	97	2.90
W1A3	6.60	6,007	123	44	6.9	165	3.10
W1B1	6.30	8,073	193	58	22	207	6.92
W1B1D	6.30	7,806	202	70	26	184	6.27
W1B2	7.20	6,471	111	51	13	86	4.64
W1B2D	6.90	6,245	124	54	16	104	4.97
W1B3	6.70	4,138	164	18	3.1	87	2.29
W1B3D	6.80	3,992	171	20	3.9	87	2.34
W1C1	6.50	6,586	223	70	20	197	6.41
W1C2	7.10	7,596	249	71	25	302	7.67
W1C3	6.60	5,589	262	62	17	157	5.62
				Cordero Mi	ne		
W2A1	6.01	6,256	62	24	51	340	73.0
W2A2	7.29	5,857	113	88	24	593	12.9
W2A3	4.23	7,477	14	89	8.8	216	8.6
W2B1	6.52	5,926	108	45	83	610	67.0
W2B1D	6.58	6,011	112	52	85	625	61.1
W2B2	7.28	3,401	38	13	57	253	140
W2B2D	7.40	3,511	44	15	61	264	132
W2B3	7.56	5,422	65	40	34	356	39.0
W2B3D	7.52	5,432	61	47	38	366	42.6
W2C1	7.06	6,274	97	51	63	509	52.1
W2C2	7.59	5,725	83	52	28	466	24.2
W2C2D	7.56	5,588	81	47	22	471	19.6
W2C3	7.65	3,087	15	6.4	36	129	212

Table 3.--Sulfur-forms analysis of spoil samples

[Concentrations are in percent by weight. Abbreviation: <, less than. Sample numbers ending in the letter "D" designate a duplicate sample]

Sample	Total sulfur	Sulfate sulfur	Sulfide	Organic sulfu
	Medicin	e Bow-Seminoe Numb	er 1 Mines	
W1A1	0.22	0.23	<0.01	0.00
W1A2	.28	.09	.10	.09
W1A3	.27	.09	.11	.07
W1B1	.19	.15	<.01	.04
W1B1D	.16	.12	<.01	.04
W1B2	.31	.08	.19	.04
W1B2D	.41	.08	.23	.10
W1B3	.42	.03	.17	.22
W1B3D	.36	.03	.13	.20
W1C1	.34	.02	.02	.30
W1C2	.17	.17	<.01	.00
W1C3	.12	.06	<.01	.06
		Cordero Mine		
W2A1	.24	.22	<.01	.02
W2A2	.18	.22	<.01	.00
W2A3	.80	.86	<.01	.00
W2B1	.30	.08	.16	.06
W2B1D	.30	.08	.14	.08
W2B2	.08	.01	.03	.04
W2B2D	.09	.02	.03	.04
W2B3	.09	.10	<.01	.00
W2C1	.06	.08	<.01	.00
W2C2	.28	.23	.01	.04
W2C2D	.13	.15	<.01	.00
W2C3	.04	.02	<.01	.02

Table 4.--Chemical analyses of

[Analyses by the U.S. Geological Survey. Concentrations in milligrams per liter (mg/L) 25 degrees Celsius;  $CaCO_3$ , calcium carbonate; <, less than. Sample numbers ending in designate the sample was split for analysis. Spoil samples used in the experiment were 25 degrees Celsius]

Sample	Specific conduc- tance (µS/cm)	рН	Cal- cium (mg/L)	Mag- nesium (mg/L)	Sodium (mg/L)	Potas- sium (mg/L)	Bicar- bonate (mg/L)	Carbo- nate (mg/L)	Alka- linity as CaCO <sub>3</sub> (mg/L)
								Ме	dicine Bow-
W1A121 W1A151 W1A221 W1A251 W1A321	3,980 3,750 4,700 4,080 4,800	6.31 6.82 7.12 7.32 6.71	260 200 400 260 370	170 160 230 170 270	510 520 520 500 490	11 11 18 14 17	460 690 800 860 720	0 0 0 0	370 560 650 700 590
W1A351 W1B121 W1B121D W1B151 W1B151D	4,250	7.14 6.74 6.66 6.81 6.78	250 370 390 280 250	200 260 280 190 180	500 640 630 560 550	13 14 16 11	810 820 790 750 740	0 0 0 0	660 670 650 610
W1B221 W1B221D W1B251 W1B251D W1B321	4,580 4,660 3,920 3,980 4,100	6.88 6.88 7.12 7.11 7.10	280 310 200 210 310	250 260 180 180 180	540 570 520 520 510	22 21 16 16 17	680 680 780 780 780	0 0 0 0	560 560 640 640 640
W1B321D W1B321S W1B351 W1B351D W1B351S	4,130 4,100 3,700 3,740 3,700	7.06 7.11 7.23 7.20 7.25	310 310 210 210 210	190 190 150 150	500 500 530 500 520	18 17 16 14 13	750 770 820 810 820	0 0 0 0	610 630 680 660 680
W1C121 W1C151 W1C221 W1C251 W1C321	5,320 4,180 5,330 4,440 4,520	6.85 7.02 7.30 7.34 6.87	410 250 390 290 380	330 190 300 200 200	610 530 630 540 530	24 18 16 13	730 780 840 870 670	0 0 0 0	600 640 690 710 550
W1C351 WWW11 <sup>1</sup> WWW12 <sup>1</sup> WWW13 <sup>1</sup>	3,930 3,410 3,420 3,450	7.05 7.44 7.42 7.47	270 130 120 130	150 140 130 130	520 530 500 530	10 10 10 9.0	770 910 900 880	0 0 0	630 740 740 720

## water from batch-mixing experiments

or micrograms per liter ( $\mu g/L$ ). Abbreviations:  $\mu S/cm$ , microsiemens per centimeter at the letter "D" designate a duplicate sample; sample numbers ending in the letter "S" collected March 1984. Batch-mixing experiments performed at a room temperature of

Sul- fate	Chlo- ride	Fluo- ride	Silica	Dis- solved solids	Alumi- num	Boron	Iron	Man- ganese	Stron- tium
(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(µg/L)	(µg/L)	(µg/L)	(μg/L)	(µg/L)
Seminoe	Number :	1 Mines							
2,000 1,500 2,500 1,900 2,500	49 34 32 27 30	0.3 .3 .3 .3	7.0 7.0 8.0 9.0 8.0	3,200 2,800 4,100 3,300 4,000	50 30 30 20 20	130 130 230 190 180	36 23 14 14 15	83 48 1,100 660 620	4,000 4,700 6,100 6,100 5,400
1,800 2,400 2,500 1,900 2,000	26 80 79 45 46	.5 .4 .4 .5	9.0 9.0 9.1 8.9 8.6	3,200 4,100 4,300 3,400 3,400	20 30 30 30 30	190 250 240 170 170	5 21 26 14 14	360 230 220 210 91	5,600 4,900 4,200 5,200 4,800
2,400 2,200 1,700 1,900 2,000	40 21 31 30 27	.3 .5 .5 .5	9.0 8.9 9.0 9.0	3,900 3,700 3,000 3,200 3,400	30 20 20 20 20	210 200 160 160 200	3 8 7 7 15	540 570 310 340 650	5,800 6,100 5,500 5,600 6,600
1,800 1,900 1,400 1,600 1,500	27 27 25 26 25	.3 .2 .3 .3	9.9 10 10 10	3,200 3,300 2,700 2,900 2,800	20 20 20 20 20 20	210 200 180 170 160	13 14 5 33 16	690 660 40 430 420	6,600 6,700 7,100 6,600 7,000
2,700 1,800 2,600 2,000 2,400	52 34 67 40 73	.5 .4 .4 .4	9.5 9.0 8.7 9.0	4,500 3,200 4,400 3,500 3,900	20 20 20 20 20 40	270 150 250 190 260	7 6 23 16 11	480 230 220 160 - 82	6,500 5,500 5,900 6,000 2,300
1,700 1,200 1,100 1,200	43 22 21 24	.3 .3 .3	10 11 11 11	3,100 2,500 2,300 2,500	20 20 20 20	160 130 160 130	77 11 10 7	59 140 120 140	2,700 9,600 9,000 9,100

Table 4.--Chemical analyses of water

Sample	Specific conduc- tance (µS/cm)	рΗ	Cal- cium (mg/L)	Mag- nesium (mg/L)	Sodium (mg/L)	Potas- sium (mg/L)	Bicar- bonate (mg/L)	Carbo- nate (mg/L)	Alka- linity as CaCO <sub>3</sub> (mg/L)
			-						Cordero
W2A121 W2A151 W2A221 W2A251 W2A321	3,730 2,670 3,880 2,910 4,040	6.62 6.78 7.22 7.20 5.64	440 270 500 330 490	120 64 220 120 270	310 260 230 230 220	30 20 9.9 8.6 8.9	610 720 910 930 160	0 0 0 0	500 590 740 760 130
W2A351 W2B121 W2B121D W2B151 W2B151D	3,320 2,740 2,670 2,000 2,010	6.16 6.87 6.87 6.93 7.05	420 220 210 130 130	150 80 78 43 44	230 310 310 250 250	7.6 25 26 18 17	390 710 720 790 820	0 0 0 0	320 580 590 650 670
W2B221 W2B221D W2B251 W2B251D W2B321	1,980 1,930 1,600 1,610 2,480	7.34 7.35 7.29 7.33 7.24	100 100 65 70 190	33 34 24 25 98	270 280 250 260 280	21 21 17 17 15	890 910 940 930 910	0 0 0 0	730 740 770 760 750
W2B321D W2B321S W2B351 W2B351D W2B351S	2,440 2,480 1,930 1,880 1,930	7.26 7.30 7.21 7.24 7.24	180 190 120 110 120	93 97 54 49 53	270 280 250 240 250	14 14 13 9.6 10	910 910 920 920 930	0 0 0 0	750 750 760 760 760
W2C121 W2C151 W2C221 W2C251 W2C321	3,410 2,260 3,720 2,620 1,810	7.25 7.22 6.43 6.74 7.55	370 180 440 260 71	150 72 190 99 27	290 260 230 230 300	17 12 30 22 22	880 910 470 650 940	0 0 0 0	720 750 380 530 770
W2C351 WWW21 <sup>1</sup> WWW22 <sup>1</sup> WWW23 <sup>1</sup>	1,530 1,320 1,320 1,330	7.47 7.20 7.28 7.19	53 •44 45 45	21 16 17 17	260 230 240 240	18 7.3 7.5 7.7	950 910 910 920	0 0 0	780 750 750 760

¹Replicate analysis of the well water (from coal aquifer) used in the batch-mixing

from batch-mixing experiments--Continued

Sul- fate (mg/L)	Chlo- ride (mg/L)	Fluo- ride (mg/L)	Silica (mg/L)	Dis- solved solids (mg/L)	Alumi- num (μg/L)	Boron (µg/L)	Iron (µg/L)	Man- ganese (µg/L)	Stron- tium (µg/L)
Mine						•			
1,700 950 1,700 940 2,400	14 13 29 19 24	0.1 .2 .6 .6	11 8.8 11 10 14	2,900 1,900 3,100 2,100 3,500	20 20 20 20 400	560 310 180 130 380	10 14 11 8 37	240 120 <1 <1 1,800	4,400 2,500 3,800 2,300 2,300
1,800 990 940 510 500	17 14 14 13 13	.3 .2 .3 .4	6.6 8.0 8.0 7.6 7.3	2,800 2,000 1,900 1,400 1,400	100 20 20 20 20 20	210 270 250 160 170	16 12 30 5 7	900 570 480 290 280	1,700 3,800 3,700 2,100 2,100
380 350 140 150 690	13 13 13 13 13	.6 .6 .7 .7	6.6 7.1 7.5 7.8 9.0	1,300 1,300 980 1,000 1,700	10 20 20 20 20	100 100 100 550 80	4 5 4 <3 10	57 58 43 42 58	1,900 2,000 1,400 1,600 2,400
730 930 350 300 340	13 14 13 13	.4 .4 .6 .6	9.0 9.0 8.8 8.5 8.7	1,800 2,000 1,300 1,200 1,300	20 20 20 20 20	80 80 90 90 80	14 12 4 3 5	49 54 35 29 35	2,300 2,400 1,500 1,400 1,500
1,200 630 2,000 1,000 210	18 15 20 15 14	.8 .7 .2 .2	16 13 6.9 6.3 8.9	2,500 1,600 3,100 2,000 1,000	20 30 30 20 10	100 100 140 120 70	11 6 19 13 <3	<1 2 190 69 11	5,100 2,500 5,100 2,900 1,800
91 2.2 1.6 2.3	13 12 12 13	.7 .9 .9	9.2 7.7 7.9 7.8	930 770 780 790	30 30 20 20	70 90 80 90	4 <3 <3 <3	11 11 11 15	1,800 860 890 890

experiment.

Table 5.--Common-constituent concentrations in

[Constituents are dissolved and concentrations are reported in milligrams per at 25 degrees Celsius; <, less than; ND, not

Well depth (feet)	Specific conduc- tance (µS/cm)	рН	Temperature		Sul- fide	Cal- cium	Mag- nesium	Sodium	Potas- sium	Bicar- bonate	Carbo- nate
							41°	54'42"	106°47'45	" (Semin	oe Number
71	5,400	6.71	11.4	0.4	ND	480	260	170	25	900	0
				•				41°5	5'36" 106	°48'01"	(Medicine
140	3,780	7.30	9.1	.6	ND	150	150	540	9.8	950	0
							44	°01'36"	105°20'4	4" (Cord	lero Mine,
110	6,000	6.78	8.9	.6	ND	580	310	680	33	600	0
							44°	01'33"	105°21'18	" (Cord	ero Mine,
188	1,478	7.22	12.5	.3	<1	48	18	250	18	940	0
			•				44	°01'13"	105°20'5	4" (Cord	ero Mine,
307	2,580	7.48	8.0	5.2	ND	250	110	190	8.2	390	0

<sup>&</sup>lt;sup>1</sup>Sample filtered through a 0.10-micrometer filter.

## water from selected wells in the mine areas

liter. Abbreviations: °C, degrees Celsius;  $\mu$ S/cm microsiemens per centimeter detected. Samples were collected June 1984]

Sul- fate	Chlo- ride		Silica	Dis- solved solids	ni-	plus	Ammonia. as	Ammonia plus organi nitrogen, as nitrogen	Phos-	
1 Mine,	spoil a	quifer	, well S	1BW-2)						
1,700	66	0.2	8.1 19.7	3,160	<0.01	<0.10	5.0	6.0	0.05	0.01
Bow Mine	e, coal	aquife	r, well	8452)						
1,400	24	.4	11 113	2,750	<.01	<.10	4.1	5.5	.01	.01
spoil a	quifer,	well M	B-26-1-P	)						
3,300	130	.3	13 113	5,640	<.01	<.10	.97	1.4	.01	.01
coal aqu	uifer, w	rell MC	-27-4-T)							
2.6	13	1.0	8.3 18.4	820	<.01	<.10	1.5	.01	.01	.01
coal aqu	uifer, w	ell MC	-27-6-P)							
1,200	1.7	.3	10 112	1,960	<.01	.51	.01	.01	.01	.01

# Table 6.--Trace-constituent concentrations in water from selected wells in the mine areas

[Constituents are dissolved, and concentrations are reported in micrograms per liter. Abbreviation: <, less than]

Date (month- day- year)	depth	Aluminum¹	Boron²	Iron¹ (	Man- ganese¹	Molyb- denum²	Selen- ium²	Stron- tium²
41°54	'42" 10	6°47'45" (Se	eminoe Nu	mber 1 Mine	e, spoil	aquife	r, well S	51BW-2)
6-25-84	71	20	100	13,000	4,800	2	<1	6,500
41°55'36" 106°48'01" (Medicine Bow Mine, coal aquifer, well 8452)								
6-26-84	140	20	12	1,100	140	<1	<1	10,000
44°01'36" 105°20'44" (Cordero Mine, spoil aquifer, well MB-26-1-P)								
6-27-84	110	<10	150	220	200	4	<1	9,100
44°01'33" 105°21'18" (Cordero Mine, coal aquifer, well MC-27-4-T)								
6-28-84	188	20	90	430	17	<1	<1	900
44	°01'13"	105°20'54"	(Cordero	Mine, coa	l aquifer	r, well	MC-27-6-	-P)
6-28-84	307	<10	100	33	18	1	7	2,800

<sup>&</sup>lt;sup>1</sup>Sample filtered through a 0.10-micrometer filter. <sup>2</sup>Sample filtered through a 0.45-micrometer filter.